PHOTOADDITION OF DIMETHYL ACETYLENEDICARBOXYLATE

TO CYCLIC ETHERS

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The free radical addition of both aliphatic and alicyclic ethers to a variety of unsaturated substrates has been a subject of recent interest^{1,2}. Jacobs and Ecke³ reported the efficient addition of tetrahydrofuran (THF) and tetrahydro-2-methylfuran to maleic anhydride but failed to observe any reaction with dimethyl acetylenedicarboxylate (DMAD). Our reinvestigation however showed that a 1:1 photoadduct is indeed formed and the findings are reported below.

DMAD in excess THF was irradiated under nitrogen with a 250-W Hanovia lamp through a pyrex filter. After 20 hours the unchanged reactants were distilled under reduced pressure, and the residue was chromatographed on silica gel to give (I) as a colourless liquid⁴. The structure (I) is based on spectroscopic data: m/e 214 (parent ion); λ_{max} (neat) 5.81, 6.04, 9.30, 9.95 μ ; δ (CC1₄) 1.7-2.3(m,4H), 3.68 and 3.73 (singlets, 6H), 5.12 (broad t, J=7 Hz, 1H), and 6.42 (s,1H).



Hydrolysis of (I) afforded a diacid (II)⁵, m.p. 112-114^o, λ_{max} (EtOH) (log ϵ) 206 (3.96), which on hydrogenation gave (III), m.p. 142-144^o. The structure of (III) was established as tetrahydrofurylsuccinic acid by spectroscopic and mixed

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m.p. comparison with an authentic sample, obtained by the hydrolysis of tetrahydrofurylsuccinic anhydride $(IV)^3$.

The mass spectral fragmentation of the unsaturated acid (II) showed no parent ion but showed a strong ion at M-18 (85%), strongly suggesting cis orientation of the carboxylic groups. This was confirmed by heating (II) at 180° when the melt showed bands in its infrared spectrum (CHCl₃) at 5.52 and 5.67 μ . The structure of the photoadduct (I) is thus secured as dimethyl tetrahydrofurylmaleate.

Irradiation of DMAD and tetrahydro-2-methylfuran gave a 1:1 adduct (V) (C-2 methyl as a singlet at g 1.42). No reaction was however observed when DMAD and 1,4-dioxan were irradiated for 24 hours.

$$\begin{array}{c} \swarrow_{0} \cdot + cH_{3} \circ oc - c \equiv c - c \circ oc H_{3} \longrightarrow \\ \downarrow & \downarrow & \downarrow \\ v_{1} & \downarrow & c \circ oc H_{3} \\ \downarrow & \downarrow & \downarrow \\ v_{11} & \downarrow & \downarrow \\ \end{array}$$

The key step in the formation of (I) is considered to involve the formation of the tetrahydrofuryl radical (VI) by the fortuitous presence of a photoinitiator in commercial THF^{1,6}. The radical (VI) adds to DMAD to give the vinyl radical (VII), which further reacts with THF to regenerate the radical (VI) and the product (I).

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References

- 1) I. Rosenthal and D. Elad, Tetrahedron, 23, 3193 (1967), and references cited therein.
- 2) R. Srinivasan and K.H. Carlough, Can. J. Chem., 45, 3209 (1967).
- 3) R.L. Jacobs and G.G. Ecke, J. Org. Chem., 28, 3036 (1963).
- 4) The photoadduct was found to be pure by t.l.c. but v.p.c. showed it ca. 90% pure.
- 5) Satisfactory elemental analyses and spectral data were obtained for all the new compounds.
- 6) 1,4-Dioxan and DMAD were found to react when the mixture was irradiated in the presence of acetone.

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